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Allow 3.10×10^{-5} if 1.6218
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as $(0.0850 - 1.62 \times 10^{-3})$ this
gives K a 3.15×10^{-5} 4 c Mol*

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$$OH - (38.2 \times 10^{-3}) \times 0.550$$

$$2.10(1) \times 10^{-2} \text{ or } 0.0210(1)$$

$$Mol H (25.0 \times 10^{-3}) \times 0.620$$

$$1.55 \times 10^{-2} \text{ or } 0.0155$$

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and Bonding Kerboodle
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water is a good conductor of*

electricity due to the presence of sodium chloride ions.

Sodium chloride ions being ionic dissociate into ions which act as charge carrier and conduct electricity.

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